THE SYNTHESIS OF 11a-CARBATHROMBOXANE As 1

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Summary. The chemically stable thromboxane analog lla-carbathromboxane synthesized from PGA2 in 12 steps. lla-Carba-TXA2 inhibits PGH2-induced aggregatelets.

Thromboxane A_2 (TXA₂)², the major product of arachidonic acid metabolism platelets, is a potent aggregatory agent and a constrictor of vascular and br muscle.² Although TXA₂ is produced by numerous tissues throughout the body, role is understood well only as it applies to platelet aggregation, since to reliable source of the labile agent (t $\frac{1}{2}$ 30-40s at 37°) is from short term in arachidonic acid or PGH₂ with human platelets.³ We report herein the synthes thromboxane A₂, in which the lla-oxygen atom of the unstable [3.1.1]bicyclic replaced by a methylene group.⁴ This chemically stable molecule was prepared that it would mimic the activity of TXA₂ itself, thus greatly simplifying the evaluation of the parent compound. The strategy of replacing an oxygen atom unit has provided chemically stable, biologically active mimics of several ot prostanoids, eg. PGH₂⁵ and prostacyclin⁶.

TXA₂ 1, IIa-CARBA-

As outlined in Figure I, PGA₂ methyl ester 15-t-butyldimethylsilyl ether to TMS-cyanohydrin 3 in 50% yield, using trimethylsilyl cyanide⁷ in dry chlore 0.25% neopentyl alcohol and 1% potassium cyanide/18-crown-6. Reduction of 3 v minum hydride afforded diol 4 (quantitative), which upon treatment with nitror 25°) yielded ring expanded β,γ -unsaturated ketone 5 (28%; ν_{max} 1705 cm⁻¹; six vinyl hydrogens; no UV absorption). There was no evidence for the formation or ring expansion regioisomer (10-keto- Δ^{11}) among the several by-products in this

11, R = H

FIGURE 1

R = H

version of β,γ -enone $\underline{5}$ to the conjugated isomer (basic alumina, tetrahydrofu stirring, 25°, 18 hr), followed by oxidation (Jones reagent, -40°) and ester methane) provided key intermediate $\underline{6}$ (40% yield from $\underline{5}$; ν_{max} 1740, 1680 cm⁻¹

Epoxidation of enone $\underline{6}$ with alkaline hydrogen peroxide afforded a mixture ketones in a ratio of 2:1 (favoring the more polar epimer, shown below to hation). After careful chromatographic separation, each epoxyketone was reducted amalgam to the corresponding β -hydroxyketone. Reduction of either β -hydroxyketone borohydride produced a 1:1 mixture of diols (epimeric at C-9), while L-Selectat -78° gave only (> 95%) the less polar isomer at C-9. Based on literature inspection of molecular models, the L-Selectride product in both cases was as configuration. The configuration at C-11 could then be readily assigned by a diol from the more polar epoxide and L-Selectride reduction (9 α ,11 α) readily butylboronate ester while diol 8 (from the minor epoxide 7 after aluminum amas Selectride reduction) did not.

Numerous attempts to form the desired 9α , 11α -oxetane by converting the rC-ll hydroxyl of diol 8 to a leaving group, followed by internal displacement ful. However, addition of 1.0 equivalent of trifluoromethanesulfonic anhydr: methylene chloride at -78° afforded, after careful isolation and chromatogral oxetane 9 (25%), accompanied by 20-30% of 13,15-diene 10 (the latter resulting alyzed dehydration of the desilylated 15-hydroxyl). Addition of triethylamine either to the triflate formation reaction mixture or at various stages during cedure, led to considerably lower yield and more by-products. This oxetane is a somewhat capricious reaction, but the yields reported above are typical. In 11α -diol (corresponding to 8) under the same conditions gave little, if any α

Ester hydrolysis under standard conditions (0.2M LiOH in 2:1 tetrahydrol 3h) afforded the desired oxetane acid $\underline{1}$, lla-carbathromboxane A₂ [R_f 0.40 in acetate/hexane/acetic acid; δ (CDCl₃; TMS) 4.1 (d, lH), 4.2-4.35 (m, lH), 4.4-5.4-6.0 ppm (m, 4H); high resolution mass spectrum (TMS derivative) M⁺ (found for C₂₇H₅₀Si₂O₄: 494.3248].

Preliminary experiments indicate that both lla-carbathromboxane A₂ $\underline{1}$ and inhibitors of PGH₂-induced aggregation of human platelets.

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